BIOENERGETICS

Adenine

Phosphate groups

Р

Р

Ribose

Р

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7.1 BIOENERGETICS AND THE ROLE OF ATP

Q.1 Define bioenergetics and discuss energy transformations in living organisms.

(Knowledge Based) (Ex Q. No 1)

Ans:

BIOENERGETICS

<u>Definition:</u> "It is the study of energy relationships and energy transformations in living organisms."

Energy Gain:

Organisms obtain energy by **metabolizing** the **food** they **eat** or **prepare**. Food **contains potential energy** in its **bonds**.

Release of Energy:

When these **bonds** are **broken down**, a large amount of kinetic energy is usually released. Some of this energy is stored in the form of potential energy in the bonds of **ATP molecules** while the rest escapes as heat.

Energy Transformations:

The potential energy stored in ATP is again transformed into kinetic energy to carry out life activities.



(MTN 2014, BWP 2014) (Ex Q. No 1)

- Q.2 Write a note on oxidation reduction reactions. (Understanding Based)
- Ans:

OXIDATION REDUCTION REACTIONS

Life Processes:

Various life processes in organisms involve constant flow of energy. This energy flow comprises the acquisition, transformation of energy and use of energy for various life processes like:

- Growth
- Movement
- Reproduction

Direct Energy Source:

For all life processes, **oxidation-reductions reactions** (redox reactions) are the **direct source of energy**.

Redox Reactions:

Redox reactions involve exchange of electrons between atoms.

Oxidation:

The loss of electrons is called oxidation.

Reduction:

The **gain of electrons** is called reduction.

Electrons as Source of Energy:

Electrons can be an energy source. It depends upon their location and arrangement in atoms. They can form two types of associations:

- Stable association
- Unstable association

Stable Association:

When electrons are present in oxygen, they make stable association with oxygen atom and are not good energy source.

Unstable Association:

If electrons are dragged away from oxygen and attached to some other atom e.g. carbon or hydrogen, they make unstable association. The try to move back to oxygen and when this happens, energy is released.

Redox Reactions in Living Organisms:

In living organisms, redox reactions involve the loss and gain of hydrogen atoms. A hydrogen atom contains one proton and one electron. It means that when a molecule loses a hydrogen atom, it actually loses an electron (oxidation). Similarly, when a molecule gains a hydrogen atom, it actually gains an electron (reduction).



Figure: Redox Reactions

Q.3 Write a note on ATP. (Knowledge Based)

(SGD 2015) (Ex Q. No 2)

Ans:

ATP: THE CELL'S ENERGY CURRENCY

Introduction:

The **major energy currency** of all **cells is a nucleotide** called adenosine triphosphate (ATP).

Discovery:

ATP was discovered in 1929 by Karl Lohmann.

Work of Fritz Lipmann:

In 1941, Fritz Lipmann proposed ATP to be the main energy-transfer molecule in the cell. He was awarded Nobel Prize.

<u>Significance:</u>

ATP is the main energy source for majority of the cellular functions like:

- Synthesis of macromolecules (DNA, RNA, proteins)
- Movement
- Transmission of nerve impulses
- Active transport
- Exocytosis
- Endocytosis

Appearance of ATP:

ATP plays a **central role as energy currency** in all organisms, it must **have appeared in the early history of life**.



Figure: Molecular Structure of ATP

Molecular Structure of ATP:

The ability of ATP to store and release energy is due to its molecular structure. Each **ATP molecule has three subunits**:

- Adenine (a double-ringed nitrogenous base)
- **Ribose** (a five-carbon sugar)
- Three phosphate groups in a linear chain

Tilde Bond:

The covalent bond connecting two phosphates is indicated by 'tilde' (~) and is a high energy bond.

Energy Release:

The energy in this bond is released as it breaks and inorganic phosphate (Pi) gets separated from ATP.

Breakdown of ATP:

The breaking of one phosphate bond releases about 7.3 Kcal (7,300 calories) per mole of ATP as follows:

 $ATP + H_2O \longrightarrow ADP + P_i + Energy (7.3 \text{ Kcal/mole})$

In common energy reactions, only the outermost of the two high-energy bonds breaks. When this happens, ATP becomes ADP (adenosine diphosphate) and one Pi is released. Breakdown of ADP:

In some cases, ADP is further broken down to AMP (adenosine monophosphate) and Pi as follows:

 $ADP + H_2O \longrightarrow AMP + P_i + Energy (7.3 \text{ Kcal/mole})$

Synthesis of ATP:

Cells constantly recycle ADP by recombining it with Pi to form ATP. The synthesis of ATP from ADP and Pi requires the expenditure of 7.3 kcal of energy per mole. This energy is obtained from the oxidation of foodstuff.

Conclusion:

ATP is generated by energy-releasing processes and is broken down by energyconsuming processes. In this way, ATP transfers energy between metabolic reactions. **Energy Storage:**

When cells use energy to build ATP from ADP, or ADP from AMP, they are storing energy as we put money in a bank.

SHORT QUESTIONS (Topic 7.1)

Q.1 What are forms of energy in living organism? (U.B)

OR

In how many forms does energy exist in living organisms? (U.B) **ENERGY FORMS IN LIVING ORGANISMS**

Ans:

In living organisms, energy exists in two forms.

Kinetic energy:

It is involved in doing any work.

Potential energy:

It is stored energy for future use. Potential energy is stored in chemical bonds and is released as kinetic energy when these bonds break.

Q.2 Define bioenergetics. (K.B)

(GRW 2012, 2015)

- Page no 202. Ans:
- 0.3 Discuss energy transformations in living organisms. (U.B)
- Ans:

TRANSFORMATIONS IN LIVING ORGANISMS

Organisms obtain energy by metabolizing the food they eat or prepare. Food contains potential energy in its bonds. When these bonds are broken down, a large amount of kinetic energy is usually released. Some of this energy is stored in the form of potential energy in the bonds of ATP molecules while the rest escapes as heat. The potential energy stored in ATP is again transformed into kinetic energy to carry out life activities.

Q.4 Recall mode of nutrition of plants, microorganisms, fungi and animals. (K.B)

Ans:

MODES OF NUTRITION

Plants and Micro-organisms:

Plants and some micro-organisms (e.g. photosynthetic bacteria and algae) prepare their own food from carbon dioxide and water in the presence of light by a process called photosynthesis.

Animals and Fungi:

Animals, fungi and many micro-organisms (non-photosynthetic bacteria and protozoans) get the prepared food.

What are redox reactions? (U.B) **Q.5**

(LHR 2013, 2014, FSD 2014, BWP 2015)

- Ans: Page no 203.
- **O.6** Can electrons be an energy source? (U.B)
- Ans: Page no 203.

- Q.7 What is the role of ATP in living organisms? (A.B)
- Ans: Page no 204.
- Q.8 What is the significance of Tilde (~) bond?
- Ans: Tilde (~) bond is a high energy covalent bond connects phosphate present in ATP. The energy in "tilde" (~) bond is released as it breaks and inorganic phosphate (Pi) gets separated from ATP.

The breaking of one phosphate bond releases about 7.3 kcal (7300 calories) per mole of ATP as follows:

```
ATP + H_2O \rightarrow ADP + Pi + Energy(7.3 \text{ kcal/mole})
```

Q.9 What is the Biochemical composition of ATP? Write down the structure of ATP? (LHR-14)

Ans: Page no 205.

- Q.10 What is meant by cell's energy currency? (K.B) (SWL 2014, MTN 2014, DGK 2015)
- Ans: Page no 205.
- Q.11 How energy is recycled in a cell? (U.B)
- Ans: Page no 206.

MULTIPLE CHOICE QUESTIONS (Topic 7.1)

All the process in cell are driven by:(K.B) 1. (A) Energy (B) Transportation (C) Anabolic reactions (D) Catabolic reaction 2. Energy stored in chemical bonds is: (K.B) (A) Potential energy (B) Kinetic energy (C) Thermal energy (D) all of these 3. Food contains energy in its bonds in the form of: (K.B) (LHR 2013) (A) Kinetic energy (B) Potential energy (C) Mechanical energy (D) All of these 4. Conversion of food into CO2 and water, transforms energy into: (U.B) (A) Chemical energy (ATP) (B) Thermal energy (C) Kinetic energy (D) All of these 5. The loss of electrons is called: (U.B) (GRW 2015, DGK 2014) (A) Oxidation (B) Reduction (C) Redox (D) Both A and B Direct source of energy in all life processes are: (K.B) 6. (A) Oxidation reactions (B) Reduction reactions (C) Redox reaction (D) Catabolic reactions Gain of electrons is called: (U.B) (A) Oxidation (B) Reduction (C) Redox (D) Both A and B ATP was discovered by: (K.B) 8. (GRW 2012, BWP 2015) (A) Fritz Lipmann (B) Karl Lohmann (C) Malvin Calvin (D) Sir Hans Krebs

	pter-7		Bioenergetics	
9.	ATP was discovered in: (K.B)	Nannally	GRW 2014, LHR 2012)	
	(A) 1923	(B) 1925		
	(C) 1927	(D) 1929		
10.	ATP is an example of: (K.B)		(LHR 2016)	
	(A) Amino acid	(B) Fatty acid		
- 0	(C) Nucleic acid	(D) Nucleotide		
n l	The ATP was proposed to be the n	nain energy transfer molecule i	n the cell by: <i>(K.B)</i>	
90	(A) Fritz Lipmann	(B) Karl Lohmann	• • •	
	(C) Rudolf Virchow	(D) Sir Hans Krebs		
12.	Each ATP molecule has subunits:	(K.B) (GRW, 2012,	, LHR 2012, SGD 2014)	
	(A) Two	(B) Three		
	(C) Four	(D) Five		
13.	Number of phosphate groups in A	TP molecule: (K.B)	(SGD 2015)	
	(A) One	(B) Two	` '	
	(C) Three	(D) Four		
14.	From which bond of ATP molecule	e energy is taken? (K.B)	(GRW 2012)	
	(A) C-N bond	(B) C-O bond		
	(C) C-H bond	(D) P-P bond		
15.	The covalent bond connected to tw	o phosphate is: <i>(K.B)</i>	(LHR 2014)	
	(A) Ratio	(B) Proportion		
	(C) Colon	(D) Tilde		
16.	How much energy is released by b	reaking of one phosphate bond	of ATP? (K.B)	
		(LHR 2012,	MTN 2014, FSD 2014)	
	(A) 7100 calories	(B) 7200 calories		
	(C) 7300 calories	(D) 7400 calories		
17.	When one inorganic phosphate is a	detached from ATP, it is conver	rted into: (K.B)	
	(A) ADP	(B) AMP	75) (CU	
	(C) GMP	(D) GTP	1000	
18.	Transformed form of potential energy from food, inside body, is stored in bonds of:			
		0 0 0 0 0 0 0 0 0 0 0 0	(U.B)	
	(A) Lipids	(B) Proteins		
	(C) Carbohydrates	(D) ATP		
19.	The high energy bond between adj	acent phosphates in ATP is: (U	(. <i>B</i>)	
NN	(A) Tilde	(B) Peptide		
90	(C) Phosphor di-ester	(D) Nucleotide		
20.	The amount of energy requires to	convert ADP to ATP is: (K.B)		
	(A) 7300 calories	(B)9000 calories		
	(7) 1000 1	(\mathbf{D}) \mathbf{Z} \mathbf{Q} 1		

7.2 PHOTOSYNTHESIS LONG QUESTIONS

What is photosynthesis? Explain intake of water and carbon dioxide. (K.B) 0.1 (GWR 2015) **PHOTOSYNTHESIS**

Ans:

Definition:

"The process of synthesis of glucose from carbon dioxide and water in the presence of sunlight and chlorophyll, with oxygen as a by-product is called as photosynthesis."

Anabolic Process:

Photosynthesis is an **important anabolic** (building) process. It comprises many coordinated biochemical reactions.

Importance:

- It is an **important component of bioenergetics** in living systems.
- It is the most important biochemical pathway and nearly all life depends on it.

Photosynthetic Organisms:

It comprises many co-ordinated bio-chemical reactions that occur in:

- Plants
- Some protists (algae) •
- Some bacteria

General Equation:

Chlorophyll $6CO_2 + 12H_2O + Light energy$ $C_6H_{12}O_6 + 6O_2 + 6H_2O$

INTAKE OF WATER AND CARBONDIOXIDE

Raw Materials:

Water and carbon dioxide are the raw materials for photosynthesis. Plants have mechanisms for the intake and transport of these raw materials.

Intake of Water:

Water, present in soil, is absorbed by root and root hairs through osmosis. This water is eventually transported to the leaves through xylem vessels.

Intake of Carbon Dioxide:

The air that enters leaf through tiny pores (stomata) reaches into the air spaces present around mesophyll cells. This air carries CO2, which gets absorbed in the thin layer of water surrounding mesophyll cells. From here, the carbon dioxide diffuses into mesophyll cells.

Role of Stomata:

Stomata cover only 1-2% of the leaf surface but they allow much air to pass through them.



Describe mechanism of photosynthesis in detail. (K.B) **MECHANISM OF PHOTOSYNTHESIS**

(DGK 2014)

Photosynthesis occurs in two phases:

- i. Light reactions
- ii. Dark reactions



i. Light Reactions:

Definition:

"The reactions which **require light** are called light reactions. They are also called **'Z-scheme.'**" Formation of High Energy Molecules:

During first phase, light energy is captured and is used to make high energy molecules:

- ATP
- NADPH

Occurrence:

These reactions, which are known as light reactions, take place on the thylakoid membranes of chloroplasts.

ii. Dark Reactions:

Definition:

"The reactions which do not require light are called light reactions. They are also called 'Calvin cycle."

Formation of Glucose:

During second phase, carbon dioxide is reduced to make glucose. In this phase, the energy from high energy molecules (ATP and NADPH) is utilized.

Occurrence:

The dark reactions take place in the stroma of the chloroplasts.

Q.3 Write a detailed note on light and dark reactions of photosynthesis. (Application Based)

(LHR 2013, 2014, 2016, GWR 2015, SWL 2015) (Ex Q. No 4)

LIGHT REACTIONS

Definition:

Ans:

"The reactions which require light are called light reactions."

Formation of High Energy Molecules:

During first phase, **light energy** is **captured** and **is used to make high energy molecules** (ATP and NADPH).

Occurrence:

These reactions, which are known as light reactions, take place on the thylakoid membranes of chloroplasts.

Mechanism:

The summary of events of light reactions is as follows:

Emission of Electrons:

When chlorophyll molecules absorb light, their energy level increases and their electrons are emitted.

Synthesis of ATP:

Electrons are passed to electron transport chain to produce ATP.

Photolysis:

Light breaks water molecule (photolysis) and oxygen is released. Hydrogen atoms of water give electrons to chlorophyll and become ions.

Reduction of NADP⁺:

The electrons of chlorophyll, after the production of ATP, and hydrogen ions of water are used for the reduction of NADP⁺ into NADPH.

Z-scheme:

The whole series of light reactions is called Z-scheme due to its Z-shaped flowchart.



Figure: Light Reactions of Photosynthesis

<u>DARK REACTIONS</u> (LHR 2013, GWR 2014, SWL 2014, DGK 2015, BWP 2015, FSD 2014, SGD 2015)

Definition:

"The reactions which do not require light are called light reactions. They are also called 'Calvin cycle'."

Discovery:

The details of dark reactions were discovered by Malvin Calvin and his colleagues at The University of California. Calvin was awarded Nobel Prize in 1961 for his work on the details of photosynthesis.

Formation of Glucose:

During second phase, carbon dioxide is reduced to make glucose. In this phase, the energy from high energy molecules (ATP and NADPH) is utilized.

Occurrence:

The dark reactions take place in the stroma of the chloroplasts.

<u>Mechanism:</u>

The summary of events of dark reactions also known as Calvin cycle is as follows:

Formation of 3-C Compounds:

CO₂ molecules are **combined** with 5-carbon compounds to form temporary 6-carbon compounds, each of which splits into two 3-carbon compounds.

Formation of 3-C Carbohydrates:

The 3-carbon compounds are reduced to 3-carbon carbohydrates by using ATP and hydrogen from NADPH. The 3-carbon carbohydrates are used to manufacture glucose.

Formation of 5-C Compounds:

The 3-carbon carbohydrates are also used to generate the original 5-carbon compounds. This step also utilizes ATP.



Q.4 Discuss role of chlorophyll and light in photosynthesis. (A.B) (SGD 2014) (Ex Q. No 3) Ans: ROLE OF CHLOROPHYLL AND LIGHT

Absorption of Sunlight Energy:

Sunlight energy is absorbed by chlorophyll. It is then converted into chemical energy, which drives the photosynthetic process.

Amount of Light Absorbed:

Only about 1% of the light falling on the leaf surface is absorbed, the rest is transmitted or reflected.

Variation in Absorption:

The light rays of different wavelengths are not only differently absorbed by photosynthetic pigments, but are also differently effective in photosynthesis. The blue and red lights carry out more photosynthesis.

<u>Pigments:</u>

Pigments are the substances that absorb visible light. Different pigments absorb light of different wavelengths (colours).

Photosystems:

Photosynthetic pigments are **organized in the form of clusters called** photosystems, in thylakoid membranes of chloroplasts.

Main Pigment:

Chlorophyll-a is the main photosynthetic pigment.

Accessory Pigments:

Chlorophyll-b and carotenoids are the accessory pigments.

Role of Chlorophyll:

Chlorophylls mainly **absorb red** and **blue lights**. Some **wavelengths not absorbed by Chlorophyll-a** are **very effectively absorbed** by **accessory pigments** and **vice versa**.

Q.5 Define limiting factor. Explain some of the limiting factors in photosynthesis. (A.B)(LHR 2015, MTN 20

Ans:

LIMITING FACTOR

Definition:

"Any environmental factor the absence or deficiency of which can decrease the rate of a metabolic reaction, is called limiting factor for that reaction."

Important Limiting Factors:

Many factors act as limiting factors for photosynthesis like:

- Light intensity
- Temperature
- Concentration of carbon dioxide
- Availability of water
- Effect of Light Intensity:

The rate of photosynthesis varies with light intensity. It decreases as the light intensity decreases and increases as the light intensity increases. However, at much higher light intensity, the rate of photosynthesis becomes constant.

Effect of Temperature:

The rate of photosynthesis decreases with decrease in temperature. It increases as temperature is increased over a limited range. But if light intensity is low, increasing the temperature has little influence on the rate of photosynthesis.

Effect of Carbon Dioxide Concentration:

As carbon dioxide concentration rises, the rate of photosynthesis goes on increasing until limited by other factors. Increase in carbon dioxide concentration beyond a certain level causes the closure of stomata, and it decreases the rate of photosynthesis.

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Q.1 Define photosynthesis. Write down its chemical equation. (K.B)

(GRW 2012, 13, 14, LHR 2013, 15, SWL 2014, MTN 2014, BWP 2015, SGD 2014, 15, RWP 2015)

Ans: Page no 209.

How CO₂ is taken into plants through leaves? (U.B)

Ans: Page no 209.

Q.2 Differentiate between dark and light reactions. (K.B) Ans: DIFFERENTIATION

(LHR 2012)

The difference between dark and light reactions are as follows:

	Light Reactions	Dark Reactions		
Phase				
It is t	he first phase of photosynthesis.	It is the second phase of photosynthesis.		
	ATP and	I NADPH		
Light	energy is captured and is used to make	Carbon dioxide is reduced to make glucose		
high-	energy molecules (ATP and NADPH)	and energy from high-energy molecules (ATP		
0	85	and NADPH) is utilized.		
	Occu	rrence		
It tak	es place on the thylakoid membranes of	It takes place in the stroma of the chloroplasts.		
chlor	oplasts.	1 1		
	Li	ght		
These	e reactions require light.	These reactions do not use light directly.		
0.3	What is $NAD^+?(K,B)$			
Ans:	NA	\mathbf{D}^+		
	Nicotinamide adenine dinucleotide (NA	\overline{AD}^+) is a coenzyme that takes electrons and		
	hydrogen ions and is thus reduced to N	NADH. One form of this coenzyme also carries		
	phosphate with it, so called NADP ⁺ .			
Q.4	What is meant by photolysis? (K.B)	(LHR 2016)		
Ans:	Page no 211.			
Q.5	Who discovered dark reactions? (K.B)			
Ans:	Page no 211.			
Q.6	6 What is the function of chlorophyll in photosynthesis? (A.B)			
Ans:	Page no 212.			
Q.7	Define photosystems. <i>(K.B)</i>	(SWL 2015)		
Ans:	Page no 212.	\mathbb{C}		
Q.8	What is meant by pigments? (K.B)	(RWP 2015)		
Ans:	Page no 212.			
Q.9	Q.9 Define limiting factor. (U.B) (MTN 2015, GRW 2015, DGK 2014, RWP 2014, 2015)			
Ans:	Page no 212.			
Q.10	vy nat is effect of light intensity on photo Page no 212	lusyntnesis: (A.B)		
Alls:	Hay the temperature offects rate of ph	actosynthesis? (A B) (I UD 2012)		
Ans	Page no 212	10105y11110515; (A.D) (LTIK 2012)		
Alls. O 12	What is the effect of carbon dioxide co	ncentration on photosynthesis? (A B)		
Q.14	what is the circle of carbon dioxide co	(GRW 2014, RWP 2015)		
Ans:	Page no 212.			

Q.13	There are more chloroplasts in the palisade mesophyll than spongy mesophyll, why? (U.B)
Ans:	MORE CHLOROPLASTS IN PALISADE MESOPHYLL
	There are more chloroplasts in the palicade mesonbyll than spongy mesonbyll because

There are more chloroplasts in the palisade mesophyll than spongy mesophyll because the palisade cells are on the upper surface and receive more light, so they contain more chloroplasts to be able to absorb more light.

Q.14 What are photosynthetic pigments? Write down the types of photosynthetic pigments. (K.B)

Ans: Pigments are the substances that absorb visible light. Different pigments absorb light of different wavelengths. There are two types of photosynthetic pigments. Main photosynthetic pigment --- Chlorophyll-a Accessory photosynthetic pigment --- Chlorophyll-b and Carotenoids. Some lights not ansorbed by chlorophyll "a" are very effectively absorbed by accessory pigments and vice versa.

Q.15 What is the significance sunlight in photosynthesis? (U.B)

Ans: Sunlight energy is absorbed by chlorophyll. It is then converted into chemical energy, which drives the photosynthetic process. Only about 1% of the light falling on the leaf surface is absorbed, the rest is reflected or transmitted.

Q.16 How wavelength of light affects the process of photosynthesis? (K.B)

- **Ans:** The sunlight of different wavelengths are not only differently absorbed by photosynthetic pigments but are also differently effective in photosynthesis. The blue and red lights carry out more photosynthesis.
- Q.17 There are more chloroplasts in the palisade mesophyll than in the spongy mesophyll. Why? (U.B)
- Ans: Because the palisade cells are on the upper surface and receive more light, so they should be able to absorb more light.

MULTIPLE CHOICE QUESTIONS (Topic 7.2)

1.	Photosynthesis is: (K.B)					
	(A) Anabolic process	(B) Catabolic process				
	(C) Metabolic process	(D) Physical process				
2.	In which process oxygen is released a	In which process oxygen is released as a byproduct?(U.B)				
		(DGK 2014, GRW 2015, BWP2014)				
	(A) Photosynthesis	(B) Respiration				
	(C) Fermentation	(D) Reproduction				
3.	Raw materials of photosynthesis are:	(K,B)				
	(A) CO_2 and O_2	(B) H_2O and CO_2				
	(C) H_2O and O_2	(D) $C_6H_{12}O_6$ and sunlight				
4.	How much surface area of leaf comp	rises of stomata? (K.B)				
	(A) 10-20 %	(B) 80-90 %				
	(C) 1-2 %	(D) 3-4%				
5.15	Water is transported to leaves throug	gh: <i>(K.B)</i>				
NNI	(A) Stomata	(B) Phloem vessels				
00	(C) Xylem vessels	(D) Lenticels				
6.	Light reactions take place on: (K.B)	(LHR 2012, 2013)				
	(A) Thylakoid membranes	(B) Stroma of chloroplasts				
	(C) Mitochondria	(D) All of these				

n

	7.	Dark reactions take place in: (K.B)	1ann//C	70000		
		(A) Stroma of chloroplasts	(B) Thylakoid membranes			
		(C) Mitochondria	(D) Cisternae			
	8.	NADPH is formed during: (U.B)	~~~			
		(A) Light reactions	(B) Dark reactions			
		(C) Krebs cycle	(D) Glycolysis			
-	2.	High energy molecules like ATP, NADPH	are synthesized in: (U.B)			
\square	11/1/	(A) Dark reaction	(B) Light reaction			
	00	(C) Calvin cycle	(D) Kreb cycle			
	10.	Light reaction occurs at/in: (K.B)				
		(A) Inter grana	(B) Stroma			
		(C) Cytoplasm	(D) Thylakoid membrane			
	11.	The break down of water molecule during	g light reactions is called: (K.B)			
		(A) Glycolysis	(B) Krebs cycle			
		(C) ETC	(D) Photolysis			
	12.	The details of dark reactions were discove	ered by:(K.B)			
		(A) Karl Lohmann	(B) Fritz Lipmann			
		(C) Malvim Calvin	(D) Louis Pasteur			
	13.	When was Calvin awarded Nobel Prize? ((K.B)			
		(A) 1960	(B) 1961			
		(C) 1962	(D) 1963			
	14.	CO2 is reduced to form glucose during: (l	J. B)			
		(A) Light reaction	(B) Z-scheme reaction			
		(C) Calvin cycle	(D) Kreb cycle			
	15.	Electrons passing through electron transp	oort chain to produce: (U.B)			
		(A) NADH	$(B) O_2$			
		(C) ATP	(D) NADPH			
	16.	Photosynthetic pigments are arranged in	the form of clusters called: (K.B)			
		(A) Carotenoids	(B) Chlorophyll-a	-ran		
		(C) Chlorophyll-b	(D) Photosystems			
	17.	Main photosynthetic pigment: (K.B)		(LHR 2016)		
		(A) Carotenoids	(B) Chlorophyll-a	700		
		(C) Chlorophyll-b	(D) All of these			
	18.	Sun light is absorbed by: (U.B)		(RWP 2015)		
		(A) Flower	(B) Stem			
		(C) Chlorophyll	(D) Roots			
	19.	Light energy is converted into chemical en	nergy during: (U.B)			
~	N	(A) Z-scheme reaction	(B) Dark reaction			
\mathcal{N}	11/11	(C) Calvin cycle	(D) Kreb cycle			
١ /	20.	Chlorophyll pigment absorbs maximum light in wavelengths of: (K.B)				
			(GRW 2013, SWL 2014	4, MTN 2015)		
		(A) Green and blue	(B) Green and red			
		(C) Green	(D) Red and blue			

- 21. Which one factor does not affect the rate of photosynthesis? (U.B) (A) Light (B) Temperature
 - (C) Humidity

- (D) Carbon dioxide
- (D) Carbon dioxide
- 22. How much percentage of light falling on the leaf surface is absorbed? (K.B) (A) 1% (B) 10 %

(D) 7%

7.3 RESPIRATION

LONG QUESTIONS

Q.1 Define respiration. Also discuss types of respiration. (K.B) (Ex Q. No 7)

RESPIRATION

Definition:

Ans:

(C) 3%

"The cellular energy-yielding process in which food is oxidized by breaking C-H bonds through oxidation-reduction reactions to produce carbon dioxide and water, is called cellular respiration."

Oxidation-Reduction Reaction:

In cellular respiration food is oxidized to carbon dioxide while oxygen is reduced to water.

TYPES OF RESPIRATION

There are two main types of respiration:

- i. Aerobic respiration
- ii. Anaerobic respiration

i. <u>Aerobic Respiration:</u>

Definition:

"The **cellular respiration occurring** in the **presence of oxygen** is called aerobic respiration."

Mechanism:

In the **presence of oxygen**, **complete oxidation** of **glucose** occurs with **maximum release of energy**. Mechanism of aerobic respiration is as follows:

<u>First Phase:</u>

In the first phase of aerobic respiration, a molecule of glucose (6-C) is broken down into two molecules of pyruvic acid (3-C).

Second Phase:

In the second phase, molecules of pyruvic acid are completely oxidized (all C-H bonds are broken) to CO₂ and water, and all energy is released.

Chemical Equation:

The overall reaction is as follows:

 $C_6H_{12}O_6 + 6O_2 \longrightarrow 6CO_2 + 6H_2O + Energy$

ii. Anaerobic Respiration (Fermentation):

Definition:

"The cellular respiration occurring in the absence of oxygen is called anaerobic respiration."

Mechanism:

In the absence of oxygen, **glucose** is **incompletely oxidized** with **less amount of energy** released. The mechanism of anaerobic respiration is as follows:

First Phase:

The first phase is exactly similar to that of aerobic respiration. A molecule of glucose is broken down into two molecules of pyruvic acid.

Second Phase:

In the second phase, **pyruvic acid** is **not completely** oxidized **due** to the **absence of oxygen**. It is **transformed into ethyl alcohol** or **lactic acid**. In this way, many of the C-H bonds are left unbroken in the products.

TYPES OF FERMENTATION

Anaerobic respiration or fermentation is further classified into:

i. Alcoholic fermentation

ii. Lactic acid fermentation

i. Alcoholic Fermentation:

In this type of anaerobic respiration, pyruvic acid is further broken down into alcohol (C_2H_5OH) and CO_2 .

Pyruvic acid — Ethyl alcohol + Carbon dioxide

Occurrence:

It occurs in:

- Bacteria
- Yeast

ii. Lactic Acid Fermentation:

In this type of anaerobic respiration, each pyruvic acid molecule is converted into lactic acid $(C_2H_6O_3)$.

Pyruvic acid _____

→ Lactic Acid

Occurrence:

It occurs in:

- Skeletal muscles of humans and other animals during extreme physical activities
- Bacteria present in milk

Q.2Discuss the importance of fermentation in detail. (A.B)(GWR 2014, DGK 2015)Ans:IMPORTANCE OF FERMENTATION

Evolution of Life:

When life evolved on Earth, the early land or water habitats did not have any supply of free oxygen (O₂). In these anaerobic conditions, early organisms respired anaerobically and got energy for their life activities.

Anaerobes:

Even today, when free oxygen is available, some organisms including some bacteria and some fungi get energy from anaerobic respiration and are called anaerobes.

<u>Strenuous Exercise:</u>

Humans can also provide energy to their skeletal muscle cells through anaerobic respiration. This happens when skeletal muscles have to work hard, (during exercise etc.) but oxygen supply cannot be increased to fulfill the demand.

Industrial Usage:

Scientists have used the fermenting abilities of fungi and bacteria for the benefit of mankind. **Examples:**

- The fermenting powers of **bacteria** are used for **making cheese** and yogurt.
- Fermentation in yeasts is used in **brewing** and **baking industries**.
- Soy sauce is made by the **fermentation** of a **fungus** *Aspergillus*.



iii. <u>Electron Transport Chain:</u>

Electron transport chain is the final step of cellular respiration. It is the transfer of electron on an electron transport chain.

Release of Electrons and H⁺:

In this step, NADH and FADH₂ release electrons and Hydrogen ions.

Role of Electron Carriers:

These electrons are taken up by a series of electron carriers.

Synthesis of ATP:

When electrons move through the series of electron carriers, they lose energy which is used to synthesize ATP.

Formation of Water:

At the end of chain, electrons and hydrogen ions combine with molecular oxygen and form water.

Q.4 Write a note on the energy budget of respiration. (K.B)

Ans:

THE ENERGY BUDGET OF RESPIRATION

ATP by NADH:

Each NADH produces 3 ATP in electron transport chain. The NADH generated in glycolysis gives 2 ATP, because one ATP is spent to transport it across the mitochondrial membrane.



ATP by FADH2:

Each FADH₂ produces 2 ATP.

Anaerobic Oxidation:

During anaerobic oxidation of glucose only 2 ATP molecules are gained as a net profit. It is because there is no Kreb's Cycle and electron transport chain in anaerobic respiration.

Net Yield:

The total net yield of ATP is 36 molecules.

Chapter-7 Q.5 Differentiate between p	ohotosynthesis and respiration. (Bioenergeti K.B) (Ex Q. No 8)	
Ans: <u>DIFFERENCES</u>	ns: <u>DIFFERENCES BETWEEN PHOTOSYNTHESIS A</u>		
Metabolism	Anabolism	Catabolism	
Energy investment / Production	Investment of light energy to store it in the form of bond	Bond energy transformed into chemical energy of	
Organisms	energy. Some bacteria, All algae, all plants	ATP. All organisms	
Site of occurrence	Chloroplasts	Cytoplasm and mitochondria	
Time of occurrence	Daytime only, in the presence of light	All the time	
Q.6 Differentiate between a	erobic and anaerobic respiration	n.	

DIFFEDENCES DETWEEN AFDODIC AND ANAFDODIC DESDIDATION

Ans:	DIFFERENCES BETWEEN AEROBIC AND ANAEROBIC RESPIRATION			
	Properties	Aerobic Respiration Anaerobic Respiration		
	Presence of Oxygen	Yes	No	
	Net profit of ATP	36	2	
Final Products		CO_2, H_2O	Lactic Acid or Ethanol + CO ₂	
	Site of occurrence	Glycolysis in cytoplasm Krebs cycle and electron transport chain in mitochondria	Cytoplasm	
	Importance	Major source of energy for all organisms	Source of energy for anaerobic organisms. Source of energy for aerobic organisms in short supply of Oxygen. Source of many products (ethanol, cheese, etc.)	

SHORT QUESTIONS (Topic 7.3)

Q.1 Define cellular respiration. (LHR 2014, 2016, BWP 2014, RWP 2014, FSD 2015)

Page no 217. Ans:

C

What is meant by Glycolysis? The breakdown of glucose into pyruvic acid is called glycolysis. Q.2 (GRW-G-15) Occurrence: Glycolysis occurs in the cytoplasm In glycolysis, glucose (6-C) molecule is broken into two molecules of Pyruvic acid (3-C). NAD⁺

> + ATP + NADH 2Pyruvic acid

ADP

Glucose

Cha	pter-7	Bioenergetics	ЭĽ
Q.3 Define aerobic respiration. Write its equ		juation. (K.B)	
Ans	Page no 217 1	(GRW 2012, 2013, 2014, SWL 2014, MTN 2015)	
0.4	What is anaerobic respiration? (K.B)	(SWL 2015)	
Ans	Page no 217.		
Q.5	What is alcoholic fermentation? (K.B)	(DGK 2015)	
Ans	Page no 218. What is lastic acid formantation? (K.B.)	(I HD 2010)	
Q.0 Ans	what is factic acid fermentation: (K.B) Page no 218	(LHR 2016)	
0.7	What is difference between aerobic and	d anaerobic respiration? (K.B)	
0.0	(LI	HR 2012, 2015, 2016, BWP 2015, FSD 2014, SGD 2015)	
Ans	<u>DIFFEREN</u>	<u>VTIATION</u>	
	The difference between aerobic and anaer	robic respiration are as follows:	
	Aerobic Respiration	Anaerobic Respiration	
	Defin	nition	
'	The cellular respiration occurring in the	The cellular respiration occurring in the	
1	presence of oxygen is called aerobic	absence of oxygen is called anaerobic	
1	espiration.	respiration.	
	Amount	of Energy	
	n the presence of oxygen, complete	In the absence of oxygen, glucose is	
	oxidation of glucose occurs with	incompletely oxidized with less amount of	
1	naximum release of energy.	energy released.	
	Number	Of AIP	
	so ATP are produced as a result of aerobic	2 ATP are produced as a result of	
08	How fermentation is important to livin	$g \text{ organisms}^2 (A B)$	
Ans:	Page no 218.		
0.9	What is the importance of fermentation	n in food industries?	
-	Scientists have used the fermenting abilit	ies of fungi and bacteria for the benefit of	
	mankind. i.e.		
	α . The fermenting powers of bacteria are	e used for making cheese and yogurt.	
	β . Fermentation in yeast is used in brew	ing and baking industries.	~
0.10	χ . Soy sauce is made by fermentation of	a fungus Aspergillus.	21
Q.10	Name the steps involved in aerobic resp	piration. (K.B)	シ
	Fage 110 219. Differentiate between photosynthesis a	nd respiration (KR)	
	Differentiate between photosynthesis a		
AU2	The difference between photosynthesis at	nd respiration are as follows:	
	Photosynthesis	Respiration are us follows.	
		ne	
It is	an anabolic reaction.	It is a catabolic reaction.	
10 10	Ene	rgy	
Lig	nt energy is stored in the form of chemical	C-H bonds in food are broken down by	
ene	·gy.	oxidation reduction reactions.	
~	Occur	rence	
It	occurs in plants, algae and some	It occurs in all organisms.	
pho	tosynthetic bacteria.		
_	Func	ction	
Ene	rgy is used to produced glucose.	Energy is released in the form of ATP.	
	BIOLO	DGY-9 222	

n

Q.12	What do you know about FAD ⁺ ?	(K.B)	(GRW 2015)		
Ans:	Flavin adenine dinucleotide (FAD ⁺) is a coenzyme like NAD ⁺ . It gets two hydrogen and				
	reduces to FADH ₂ .) is a coefficience interview . It ge	is two nytrogen and		
0.13	Why it is incorrect to say that	energy releasing step of resp	oiration is electron		
Ľ	transport chain? (U.B)				
Ans:	ENERGY RELEASE	NG STEP OF RESPIRATION			
NNN	It is incorrect to say that energy re	leasing step of respiration is elec	etron transport chain		
90	because energy is released in glyc	olysis and Kreb's cycle in the	form of NADH and		
	FADH ₂ . Electron transport chain the	ransforms the energy present in	these compounds to		
0.14	ATP.				
Q.14	How many energy molecules are pl	roduced in aerobic and anaerobic	c respiration? (K.B)		
Ans:	<u>AIPI</u> Aarabia Pospiration:	PRODUCTION			
	36 ATP molecules are produced as	a result of aerobic respiration			
	Anaerobic Respiration:	a result of acrobic respiration.			
	2 ATP molecules are produced as a	result of anaerobic respiration.			
	MULTIPLE CHOICI	E QUESTIONS (Topic	7.3)		
1.	From which bond of food energy	is taken? (K.B)			
	(A) P-P bonds	(B) C-H bonds			
	(C) C-N bonds	(D) C-O bonds			
2.	The greatest fuel of energy of cells	ular respiration is: (U.B)	(RWP 2014)		
	(A) Glucose	(B) Proteins			
-	(C) Amino acid	(D) Lipids			
3.	Oxidation of glucose depends upo	n: (U.B)			
	(A) Availability of CO_2	(B) Availability of O_2			
4	(C) Availability of H_2O	(D) Availability of A I P $di = d 4 a a (K R)$			
4.	(A) Carbon diavida	$(\mathbf{P}) \mathbf{H}_{\mathbf{O}}$			
	(C) Both A and B	(D) Ghucose	- 199		
5.	The most common fuel used by ce	Il to get energy from cellular reg	spiration is: (KB)		
01	(A) Glucose	(B) Oxygen			
	(C) Carbon dioxide	(D) Food	1 Caso		
6.	Through which process organism	gets energy? (K.B)	(SGD 2015)		
	(A) Photosynthesis	(B) Respiration			
	(C) Transpiration	(D) Evaporation			
7.	Which of the following reaction is	associated to aerobic respiratio	n? (U.B)		
-	(A) Glycolysis	(B) Calvin cycle			
NA/	(C) Electron transport chain	(D) Alcoholic fermentat	ion		
UN V	Which of the following is common	in both aerobic and anaerobic	respiration? (U.B)		
0	(A) Glycolysis	(B) Kerb cycle			
0	(C) Electron transport chain	(D) Alcoholic termentat	ion		
у.	Accononic termentation occurs in:	$(\mathbf{A}, \boldsymbol{\mathcal{B}})$ (B) Vecato			
	(A) Daviella (C) Viruses	(D) Humans			
		(D) Huilialis			

0

10.	Whose fermenting powers are used for making cheese and yogurt? (A.B) (SGD 2014)		
	(A) Bacteria	(B) Viruses	
	(C) Fungi	(D) Algae	- 0
11.	In which phase of respiration glu	cose molecule is broken	n into two molecules of
	pyruvic acid? (K.B)		
	(A) Glycolysis	(B) Krebs cycle	
- 10	(C) Electron transport chain	(D) Light reaction	
12.	Soy sauce is made through the ferm	entation by a fungus: (A.	<i>B</i>)
90	(A) Rhizopus	(B) Penicillium	
	(C) Aspergillus	(D) Puccinia	
13.	How many stages are present in aer	obic respiration? (K.B)	
	(A) 1	(B) 2	
	(C) 3	(D) 4	
14.	Yogurt and cheese is made by using	the fermentative abilitie	es of: (U.B)
	(a) Bacteria		(b) Plants
	(c) Fungi	(d) Yeast	
15.	Process of glycolysis is found in: (K.)	B)	(GRW 2014, LHR 2015)
	(A) Ribosomes	(B) Cytoplasm	
	(C) Golgi complex	(D) Vacuole	
16.	In electron transport chain each NA	DH produces: (K.B)	
	(A) 2 ATP	(B) 3 ATP	
	(C) 4 ATP	(D) 5 ATP	
17.	In which step of respiration CO ₂ is	produced? (K.B)	(GRW 2013)
	(A) Glycolysis	(B) Krebs cycle	
	(C) Electron transport chain	(D) All of these	
18.	How many molecules of CO2 are pro	duced when Krebs cycle	operates once? (K.B)
			(LHR 2015, MTN 2015)
	(A) 6	(B) 3	
	(C) 2	(D) 1	
19.	How many ATP molecules are gener	rated in aerobic respirat	ion? (K.B)
			(GRW 2009, LHR 2014)
	(A) 2	(B) 24 (D) 26	
• •	(C) 34	(D) 36	NICO LOS
20.	In cell, aerobic respiration sites are:	(K.B)	(SGD 2015)
	(A) Golgi bodies	(B) Mitochondria	
	(C) Ribosomes	(D) Chloroplast	
21.	Krebs cycle and electron transport	chain occur in: (K.B)	
	(A) Cytoplasm	(B) Nucleus	
	(C) Mitochondria	(D) Ribosomes	
22.	How many ATP molecules are ut	tilized for passage of e	ach Pyruvic acid inside
WI.	mitochondria? (U.B)	(m) -	
50	(A) 1	(B) 2	
	(C) 3	(D) 6	



7.2 PHOTOSYNTHESIS

1	А	6	А	11	D	16	D	21	С
2	А	7	А	12	С	17	В	22	А
3	В	8	А	13	В	18	С		
4	С	9	В	14	С	19	А		
5	С	10	В	15	С	20	D		

7.3 RESPIRATION





UNDERSTANDING THE CONCEPTS

- 1. How would you define bioenergetics while relating it to the oxidation-reduction reactions in living systems? (U.B)
- Ans: See the LQ.1 and 2 of (Topic 7.1)
- 2. Interpret that ATP is the chief energy currency of all cells. (U.B)
- Ans: See the LQ.3 of (Topic 7.1)
- 3. What is the role of chlorophyll and light in photosynthesis? (A.B)
- Ans: See the LQ.4 of (Topic 7.2)
- 4. Outline the processes involved in photosynthesis. *(K.B)*
- Ans: See the LQ.2 of (Topic 7.3)
- 5. State how the varying light intensity, carbon dioxide concentration and temperature affect the rate of photosynthesis? (*A.B*)
- Ans: See the LQ.5 of (Topic 7.2)
- 6. Outline the mechanism of respiration while defining glycolysis, Krebs cycle, and electron transport chain. (K.B)
- Ans: See the LQ.3 of (Topic 7.3)
- 7. Draw a comparison of aerobic and anaerobic respiration. (K.B)
- Ans: See the LQ.6 of (Topic 7.3)
- 8. How will you compare respiration and photosynthesis? (K.B)
- Ans: See the LQ.5 of (Topic 7.3)

SHORT QUESTIONS

1. Why is it said that all life forms are dependent on photosynthesis? (U.B)

Ans:

Ans: 3.

ALL LIFE FORMS ARE DEPENDENT ON PHOTOSYNTHESIS

All life forms on the earth ultimately depend on each other for their energy requirements.

<u>Autotrophic Organisms:</u>

Autotrophic organisms such as plants are capable of producing their own food by the process of photosynthesis. Autotrophic organisms lie at the base of food chain which implies that all life forms are dependent on the process of photosynthesis, directly or indirectly.

Heterotrophic Organisms:

All other organisms are heterotrophs, i.e. depend on other organisms for their food requirements. Food is manufactured by autotrophic organisms via photosynthesis, which is used by themselves and also animals that feed on them.

2. What structures and phenomena are involved in the intake of carbon dioxide and water by plants? (K.B)

See the LQ.1 of (Topic 7.2)

- In what ways the respiratory energy is used in the body of organisms? (K.B)
- Ans: See the SQ.7 of (Topic 7.1)
- 4. What is the importance of anaerobic respiration? (A.B)
- Ans: See the LQ.2 of (Topic 7.3)





(D) Light Reactions of Photosynthesis Instructions:

• Trace the patterns and marks the labels





(F) Energy Chart of Respiration **Instructions:**

Trace the patterns and mark the labels •



×	Chap	ter-7	Bioen	ergetics						
CUT HERE										
1	Time: Q.1	40 min Four possible answers A, B, C and D to	N each question are given, mark th	larks: 25 e correct						
Í	1.	answer. Food contains energy in its bonds in the f	orm of: <i>(K.B)</i>	(6×1=6)						
W		(A) Kinetic energy	(B) Potential energy							
	NN/	(C) Mechanical energy	(D) All of these							
	2.	ATP is an example of: (K.B)								
		(A) Amino acid	(B) Fatty acid							
		(C) Nucleic acid	(D) Nucleotide							
1	3.	When one inorganic phosphate is detache	ed from ATP, it is converted into: (K.B)						
I		(A) ADP	(B) AMP							
I		(C) CMP	(D) GMP							
I	4.	Light reactions take place on: (K.B)								
I		(A) Thylakoid membranes	(B) Stroma of chloroplasts							
1		(C) Mitochondria	(D) Inner mitochondrial membrane							
	5.	Which one factor does not affect the rate	of photosynthesis? (U.B)							
I		(A) Light	(B) Temperature							
I		(C) Humidity	(D) Carbon dioxide concentration							
I	6.	How many stages are present in aerobic r	respiration? (K.B)							
I		(A) 1	(B) 2							
1		(C) 3	(D) 4							
I	Q.2	Give short answers to following questions		(5×2=10)						
I	1.	Define bioenergetics. (K.B)	213	C(0)UUU						
1	2.	Draw structure of ATP. (K.B)	1-nrall(Q)	000						
I	3.	How carbon dioxide is taken into plants through	ough leaves? (A.B)							
1	4.									
1	5.	What is effect of light intensity on photosyn	thesis? (U.B)							
W	Q.3	Answer the following Questions in detail.		(5+4=9)						
	(a)	Write a note on light reactions. (K.B)		(5)						
	(b)	Describe mechanism of respiration. (K.B)		(4)						
I	Note:									
I		Parents or guardians can conduct this test in their supervision in order to check the skill of students.								