

# SHORT QUESTIONS

**1.1 Name several repetitive phenomenon occurring in nature which could serve as reasonable time standard.**

**Ans.** Any natural phenomenon that repeats itself after exactly same time interval can be used as a measure of time. The repetitive phenomenon could serve as reasonable time standard, occurring in nature are as follows:

1. **Sun:** Sun served as reasonable time standard because sunset and sunrises gives the information of time.
2. **Moon:** Moon is also reasonable time standard because it gives the information of time.
3. **Weather:** Changing of weather can also give information about time.
4. Rotation of Earth on its axis.
5. Rotation of Earth around the sun.
6. Oscillation of a simple pendulum.

**1.2 Give draw backs to use the period of a pendulum as a time standard.**

**Ans.** As we know that the time period of a simple pendulum depends upon the length and value of  $g$  at any place. Since

$$T = 2\pi \sqrt{\frac{l}{g}}$$

- (i) It is clear that time period of a simple pendulum depends upon the value of  $g$  which is different at different places. So a pendulum of same length may have different time period at difference places. So period of pendulum cannot be taken as standard for measuring time.
- (ii) **Friction:** Time period of a simple pendulum changes due to air resistance.
- (iii) **Temperature:** In summer due to increase in temperature, length of simple pendulum changes so time period changes.

**1.3 Why do we find it useful to have two units for the amount of substance, the kilogram and the mole?**

**Ans.** It is very useful to have two units for the amount of substance i.e., kilogram and mole. If we want to consider a specific amounts of mass without considering number of microscopic atoms present in it, it is useful to use **kilogram**. Because one kilogram of different substances contains different number of molecules. While if we want to consider a fixed number of atoms present in it then it is useful to use **mole**. Because one mole of any substance contains the same number of atoms or molecules.

**1.4 Three student's measured the length of a needle with a scale on which minimum division is 1 mm and recorded as (i) 0.2145m (ii) 0.21m (iii) 0.214m. Which record is correct and why?**

**Ans.** In these records (iii) 0.214 m is more correct than the other records because the least count of a scale is 1 mm which can be written as 0.001 m. So according to this figure, the student measure that type of record is correct.

**1.5** An old saying is that “A chain is only as strong as its weakest link”. What analogue statement can you make regarding experimental data used in computation?

**Ans.** The analogous statement regarding experimental data used in computation is “A result obtained from an experimental data used in computation is only as accurate as its least accurate reading”.

**1.6** The period of simple pendulum is measured by a stopwatch. What type of errors are possible in the time period?

**Ans.** When the period of a simple pendulum is measured by a stopwatch, the following types of errors are possible:

1. **Systematic Error:** The error due to the fault in the measuring instrument is called systematic error i.e., zero error.
2. **Personal Error:** The error due to the faulty procedure of an observer is called personal error.

**1.7** Does dimensional analysis give any information on constant of proportionality that may appear in an algebraic expression? Explain.

**Ans.** Dimensional analysis does not give any information about the constant of proportionality or dimensionless constant. For example

$$v = \text{Constant} \times \sqrt{\frac{E}{\rho}}$$

The numerical value of this constant cannot be determined by dimensional analysis.

**1.8** Write the dimension of:

- (i) Pressure                      (ii) Density

**Ans.** (i) **Dimensions of Pressure:**

$$\text{As } P = \frac{F}{A} = \frac{ma}{A}$$

$$\text{Unit of } P = \frac{\text{kg ms}^{-2}}{\text{m}^2} = \text{kg m}^{-1}\text{s}^{-2}$$

$$\Rightarrow [P] = [ML^{-1}T^{-2}]$$

(ii) **Dimensions of Density:**

$$\text{As } \text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

$$\text{Unit of density} = \frac{\text{kg}}{\text{m}^3} = \text{kg m}^{-3}$$

$$[\text{Density}] = [ML^{-3}]$$

**1.9** The wavelength  $\lambda$  of a wave depends on the speed  $v$  of the wave and its frequency  $f$ . Knowing that:

$$[\lambda] = [L], [v] = [LT^{-1}] \text{ and } [f] = [T^{-1}]$$

Decide which of the following is correct,  $f = v\lambda$  or  $f = \frac{v}{\lambda}$ .

**Ans.** In 1<sup>st</sup> case if  $f = v\lambda$  where  $f$  is frequency. Its dimension is  $[T^{-1}]$ ,  $v$  is speed, its dimensions are  $[LT^{-1}]$ .

$\lambda$  is the wavelength, its dimension is  $[L]$ .

$$\text{So,} \quad [T^{-1}] = [LT^{-1}][L]$$

$$[T^{-1}] = [L^2T^{-1}]$$

Hence the equation  $f = v\lambda$  is not dimensionally correct because left hand side dimension is not equal to right hand side dimension.

In second case

$$f = \frac{v}{\lambda}$$

$$\text{So} \quad [T^{-1}] = \frac{[LT^{-1}]}{[L]}$$

$$[T^{-1}] = [T^{-1}]$$

Hence the equation  $f = \frac{v}{\lambda}$  is dimensionally correct because left hand side dimensions is equal to right hand side dimension.

**Ans.** In 1<sup>st</sup> case if  $f = v\lambda$  where  $f$  is frequency. Its dimension is  $[T^{-1}]$ ,  $v$  is speed, its dimensions are  $[LT^{-1}]$ .

$\lambda$  is the wavelength, its dimension is  $[L]$ .

$$\text{So,} \quad [T^{-1}] = [LT^{-1}][L]$$

$$[T^{-1}] = [L^2T^{-1}]$$

Hence the equation  $f = v\lambda$  is not dimensionally correct because left hand side dimension is not equal to right hand side dimension.

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Hence the equation  $f = \frac{v}{\lambda}$  is dimensionally correct because left hand side dimensions is equal to right hand side dimension.