

CHAPTER

4

GROUP VA AND GROUP VIA ELEMENTS

MULTIPLE CHOICE QUESTIONS

- 8.** When red phosphorous is heated with HNO_3 it forms:
(a) H_3PO_4 (b) HPO_2
(c) H_2PO_3 (d) H_3PO_3

9. Which of the following is used at the tips of match stick:
(a) $\text{K}_2\text{Cr}_2\text{O}_7 + \text{S} + \text{White P}$ (b) $\text{K}_2\text{Cr}_2\text{O}_7 + \text{K} + \text{S}$
(c) S and K (d) Sb_2S_3

10. What is the number of electrons present in the valence shell of P in PCl_3 :
(a) 4 (b) 6
(c) 8 (d) 2

11. Phosphorous shows oxidation state (+3) in which of the following:
(a) $\text{H}_4\text{P}_2\text{O}_7$ (b) PO_4^{-3}
(c) H_3PO_3 (d) H_3PO_4

12. Which element is the most abundant in the earth's crust:
(a) Fe (b) O
(c) Si (d) C

13. Which catalyst is used in contact process:
(a) Fe_2O_3 (b) V_2O_5
(c) SO_3 (d) Ag_2O

14. The brown gas formed when metal reduces HNO_3 is:
(a) N_2O_5 (b) N_2O_3
(c) NO_2 (d) NO

15. Which of the following specie has the maximum number of unpaired electrons:
(a) O_2^{-1} (b) O_2^{+2}
(c) O_2^{-2} (d) O_2

16. The contact process for the manufacturing of H_2SO_4 was developed by:
(a) Jaber-Bin-Hayan (b) Knietsch
(c) Al-Khwarzmi (d) Mendleeve

17. In which compound of nitrogen, the oxidation state of N is (+1):
(a) N_2O (b) NO
(c) NO_2 (d) N_2O_4

18. **FeSO₄** forms brown ring with:

 - (a) N₂O₄
 - (b) NO
 - (c) NO₂
 - (d) None of these

19. "Lead" in lead pencil is:

 - (a) Bone black
 - (b) Graphite and clay
 - (c) Lead oxide
 - (d) Lead peroxide

20. Formula of Gibbsite is:

 - (a) Al₂O₃
 - (b) Al₂O₃ . H₂O
 - (c) Al₂O₃ . 2H₂O
 - (d) Al₂O₃ . 3H₂O

answers

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|------------|-----|------------|-----|------------|-----|------------|-----|------------|-----|
| 1. | (b) | 2. | (c) | 3. | (d) | 4. | (b) | 5. | (b) |
| 6. | (a) | 7. | (c) | 8. | (a) | 9. | (a) | 10. | (c) |
| 11. | (c) | 12. | (b) | 13. | (b) | 14. | (c) | 15. | (d) |
| 16. | (b) | 17. | (a) | 18. | (b) | 19. | (b) | 20. | (d) |